## TVSSC - COMMON REVISION EXAMINATION - (2019 - 2020) SCIENCE (086) MARKING SCHEME

SECTION – A		
1	concave mirror, to get the parallel beam of light.	1 mark
2	Volt is the electrical unit of voltage (or) potential difference	1 mark
3	The advantage of solenoid over an ordinary coil is that inside a	1 mark
	solenoid a uniform magnetic field is produced	
4(a)	artificial dialysis	1 mark
4(b)	iii) asthma	1 mark
4(c)	kidney failure	1 mark
4(d)	GFR value of 90mL/min or higher indicates that glomerulus is	4
	functioning properly. GFR values less than 90mL/min indicates the	1 mark
	kidney damage.	
5(a)	A belongs to group - 1 and G belongs to group - 17	1 mark
5(b)	H belongs to noble gas element	1 mark
5(c)	A will have largest atomic radius	1 mark
5(d)	D is likely to be a metalloid or semi metal	1 mark
6	c) Cinematography (or) a) Red colour is least scattered	1 mark
7	(b) 0.025 V	1 mark
8	(a) 6.5 – 7.5	1 mark
9	(d) (ii) and (iv)	1 mark
	OR	
	(b) neon	
10	(c) Conc.HCI : Conc.HNO <sub>3</sub>	1 mark
	3 : 1	
11	b) Hydro electricity	1 mark
12	(b) 10%	1 morts
	OR	1 mark
	(a) khadins	
13	c) If assertion is true but reason is false.	1 mark
14	(d) A is false but R is true.	1 mark

SECTION - B		
15	(i) The colour of ferrous sulphate crystals changes from green to white	0.5 Mark
	then turn into brown.  The smell of burning sulphur is observed.	0.5 Mark
	(ii) Thermal decomposition reaction	0.5 Mark
	(iii) 2FeSO <sub>4</sub> Fe <sub>2</sub> O <sub>3</sub> + SO <sub>2</sub> + SO <sub>3</sub>	1 Mark
	The products are - Ferric Oxide, sulphur dioxide and sulphur trioxide.	0.5 Mark
16	Al <sub>2</sub> O <sub>3</sub> + 2NaOH → 2NaAlO <sub>2</sub> + H <sub>2</sub> O	0.5 Mark
	Al <sub>2</sub> O <sub>3</sub> + 6HCl → 2AlCl <sub>3</sub> + 3H <sub>2</sub> O	0.5 Mark
	$ZnO + 2NaOH \longrightarrow Na_2ZnO_2 + H_2O$	0.5 Mark
	$ZnO + 2HCI \longrightarrow ZnCl_2 + H_2O$	0.5 Mark
	There are called amphoteric Oxides.	1 Mark
	OR	
	(i) Ca + 2H <sub>2</sub> O → Ca(OH) <sub>2</sub> + H <sub>2</sub>	1 Mark
	(ii) 2HgS + 3O <sub>2</sub> → 2HgO + 2SO <sub>2</sub>	1 Mark
	(iii) 3MnO <sub>2</sub> + 4Al → 2Al <sub>2</sub> O <sub>3</sub> + 3Mn + Heat	1 Mark
17	(i) K	0.5 Mark
	(ii) Be and Ca, as both have same number of valence electrons (ie)., 2.	1 Mark
	(iii) K <sup>+</sup> + X <sup>-</sup> → K X	1 Mark
	Nature of the compound is ionic.	0.5 Mark
18	- It takes place in the small intestine. Fats are present in the intestine in	
	the form of large globules.	0.5 mark
	- Bile salts break them down into smaller globules increasing efficiency	1 mark
	of enzyme action.	
	- Pancreatic juice has enzyme lipase which breaks down emulsified fat.	1 mark
	- Enzymes secreted by the walls of small intestine finally converts fats into fatty acid and glycerol	0.5 mark
	OR	
	(a) Adrenaline increases the heart beat and breathing rate which results in the supply of more oxygen to muscles. It reduces the blood supply to the digestive system and skin and diverts it to the skeletal muscles. All these responses together enable us to deal with emergency situation.	1 mark
	(b) If growth hormone is secreted in excess it leads to gigantism, while if secreted less leads to dwarfism in childhood.	1 mark
	(c)Patients suffering from diabetes have high blood sugar level as insulin is not secreted by pancreas which helps to lower blood sugar level.	1 mark

19	(a)Removal of stamens of a bisexual flower will not affect pollination as	
	its pistil is intact.	1 mark
	Formation of fruit will take place as transfer of pollen grains from the	1 mark
	anther of another flower to the stigma will take place causing cross	
	pollination.	
	(b) It transfers glucose and oxygen form mother's blood to the foetus.	0.5 mark
	It removes the waste generated by the foetus to the mother's blood.	0.5 mark
20	(a) Rr and Rr	1 mark
	(b) Red colour flower is dominant and white colour flower is recessive.	1 mark
	(c) Monohybrid cross, phenotypic ratio is 3:1	1 mark
21	Any food chain that could exist in a grassland ecosystem with plants at the first trophic level.	1 mark
	$Sun \xrightarrow{1\%} Plant \xrightarrow{10\%} Deer \xrightarrow{10\%} Lion$	2 mark
	30,000J $300J$ $30J$ $3J$	
22	<ul> <li>i) In a metal, large number of free electrons are found. These electrons are in constant random motion that means they are moving here and there. Thus, current generated in one direction is cancelled out by current generated in opposite direction by these randomly moving electrons. So, net current in a metal is zero.</li> <li>ii) The resistance of an ideal ammeter is zero.</li> <li>OR</li> <li>i) V = 1.2V</li> </ul>	2 Mark  1 Mark  0.5 mark
	$I = 2\mu A = 2 \times 10^{-6} A$	0.5 mark
	$R = \frac{V}{I} = \frac{1.2 \times 10^6}{2} = \frac{12 \times 10^5}{2} = 6 \times 10^5 \Omega$	1 Mark
	ii) Parallel	1 Mark
23	Near point is 1 m = 100 cm	0.5 mark
	Here $u = -25$ cm,	0.5 mark
	v= -100 cm, f = ? $\frac{1}{f} = \frac{1}{v} - \frac{1}{u} = \frac{1}{-100} - \frac{1}{-25} = \frac{3}{100} ; \frac{1}{f} = \frac{3}{100}$	0.5 mark
	$P = \frac{3}{100} cm^{-1} = \frac{3}{100} \times 100m^{-1} = 3D$	0.5 mark

	(a) Near point of a Hypermetropic eye  (b) Hypermetropic eye  (c) Correction for Hypermetropic eye	1 Mark
24	$P = \frac{E}{t}$ ;	0.5 Mark
	$E = P \times t = 2KW \times 3600  s = 7200k \times \frac{J}{s} \times s$	0.5 Mark
	∴ 18 KJ of heat is liberated on burning 1 g of coal.	1 Mark
	∴ 7200 KJ of heat is liberated by burning = $\frac{7200 \times 1}{18}$ = 400 g of coal.	
	ii) The fresh piece of wood contains moisture. The heat given is used to remove moisture and, therefore it is difficult to burn it.	1 Mark
25	(i) CH <sub>3</sub> CH <sub>2</sub> OH	1 Mark
	(ii) CH <sub>3</sub> CH <sub>2</sub> OH + CH <sub>3</sub> COOH Conc. H <sub>2</sub> SO <sub>4</sub> CH <sub>3</sub> COOC <sub>2</sub> H <sub>5</sub> + H <sub>2</sub> O	1 Mark
	(iii) CH <sub>3</sub> COOC <sub>2</sub> H <sub>5</sub> + NaOH	1 Mark
	(iv) CH <sub>3</sub> COOH + NaHCO <sub>3</sub> → CH <sub>3</sub> COONa + CO <sub>2</sub> + H <sub>2</sub> O	1 Mark
	(v) 2CH₃CH₂OH + 2Na   → 2CH₃CH₂ONa + H₂  OR	1 Mark
	(i) $CH_3CH_2OH$ $Conc. H_2SO_4$ $CH_2 = CH_2 + H_2O$	1 Mark
	Ethanol Ethene Conc. H <sub>2</sub> SO <sub>4</sub> acts as dehydrating agent.	1 Mark
	$\begin{array}{c} \text{Ni} \\ \text{CH}_2\text{=CH}_2 + \text{H}_2 & \longrightarrow & \text{CH}_3 - \text{CH}_3 \\ \text{Ethene} & \text{Ethane} \end{array}$	1 Mark

	(ii) It cannot lose four electrons, because high energy is needed to remove four electrons.  It cannot gain four electrons, because 6 protons cannot hold 10	1 Mark
	electrons.	1 Mark
26	(i) Tap water contains ions which makes it a good conductor whereas	1 Mark
	distilled water does not contain any ions.	
	(ii) Dry HCl gas does not dissociate into ions, so it has no effect on the	4.841
	litmus. Hydrochloric acid form ions. So it turns blue litmus red.	1 Mark
	(iii) Baking soda prevents the formation of lactic acid when milk turns	1 Mark
	sour.	
	(iv) Acid is added to water slowly because the reaction is highly	1 Mark
	exothermic. If water is added to acid, then glass container may	
	break due to lot of heat evolved.	
	(v) NH <sub>3</sub> dissolves in H <sub>2</sub> O forming NH <sub>4</sub> OH, therefore it acts as base.	1 Mark
	NH <sub>3</sub> + H <sub>2</sub> O → NH <sub>4</sub> OH.	
27		_
	Refer NCERT text book : Pg.268, 269 Each point carries 1 mark	5 mark
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28	a) b) P – Hypothalamus - controls body temperature	
	Q – Pituitary - controls growth	
	R – Medulla - controls involuntary actions	
	like blood pressure, salivation etc.	
		5 mark
	S – Cerebellum - controls precise voluntary actions,	(1 mark each)
	balance and posture.	
	T – Cerebrum - sensing, memory storage and conscious	
	behaviour.	
	OR	
	i) produce sperm male sex hormone testosterone.	
	ii) secrete fluid that provide nutrition to the sperms.	
	iii) sperms are carried through vas deferens which write with a tube	5 mark
	coming from urinary bladder.	(1 mark each)
	iv) tube that carries urine from kidney to the urinary bladder.	
	v) add secretions so that sperms are in a fluid that facilitates their	
	transport and also provide nutrition.	

29	(i) $u = -35 cm$	
	f = -20cm	0.5 mark
	v = ?	
	h = 10 cm	
	$h^1 = ?$	
	$\frac{1}{f} = \frac{1}{u} + \frac{1}{v}$	0.5 mark
	$\frac{1}{v} = \frac{1}{f} - \frac{1}{u} = \frac{1}{-20} - \frac{1}{-35}$	O. F. mag rile
	$\frac{1}{v} = \frac{1}{-20} + \frac{1}{35} = -0.05 + 0.03$	0.5 mark
	$\frac{1}{v} = -0.02$	
	v = -50 cm	0.5 mark
	$h^1 = -\frac{vh}{u} = -\frac{(-50) \times 10}{-35}$	0.5 mark
	$= -\frac{500}{35} = -14.3 \ cm$	
	Nature: Real and inverted.	0.5 mark
	ii) $m = -3$	oro man
	$m=-\frac{v}{u}; \ 3=-\frac{v}{u};$	0.5 mark
	v = 3u;	0.5 mark
	u = -20cm	
	$v = 3 \times -20$	0.5 mark
	v = -60 cm	0.5 mark
30	i) $\frac{1}{R'} = \frac{1}{R_1} + \frac{1}{R_2}$	0.5 Mark
	$\frac{1}{R'} = \frac{1}{10} + \frac{1}{40} = \frac{5}{40}$	
	$R'=8\Omega$	0.5 Mark
	$\frac{1}{R''} = \frac{1}{R_3} + \frac{1}{R_4} + \frac{1}{R_5}$	0.5 Mark
	$\frac{1}{R''} = \frac{1}{30} + \frac{1}{20} + \frac{1}{60} = \frac{6}{60}$	0.514
	$R'' = 10\Omega$	0.5 Mark
	$R = R' + R'' = 18\Omega$	0.5 Mark
	Ohm's law $I = \frac{V}{R} = \frac{12V}{18\Omega} = 0.67A$	0.5 Mark

